

Non-Thermal Plasma-Assisted VOC Decomposition: Influence of Catalyst Type, Electron Energy, and Gas Residence Time



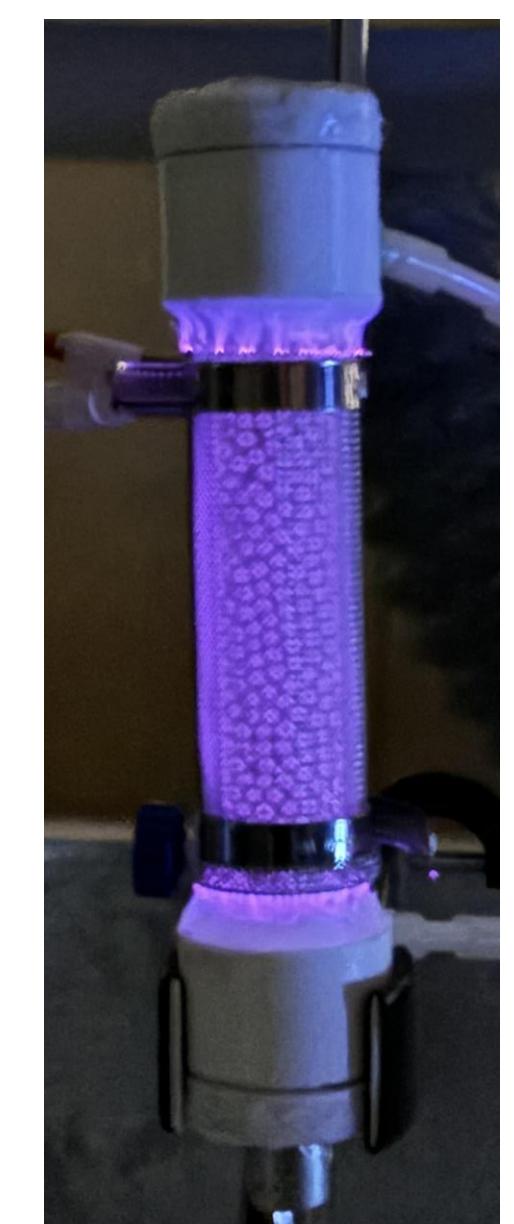
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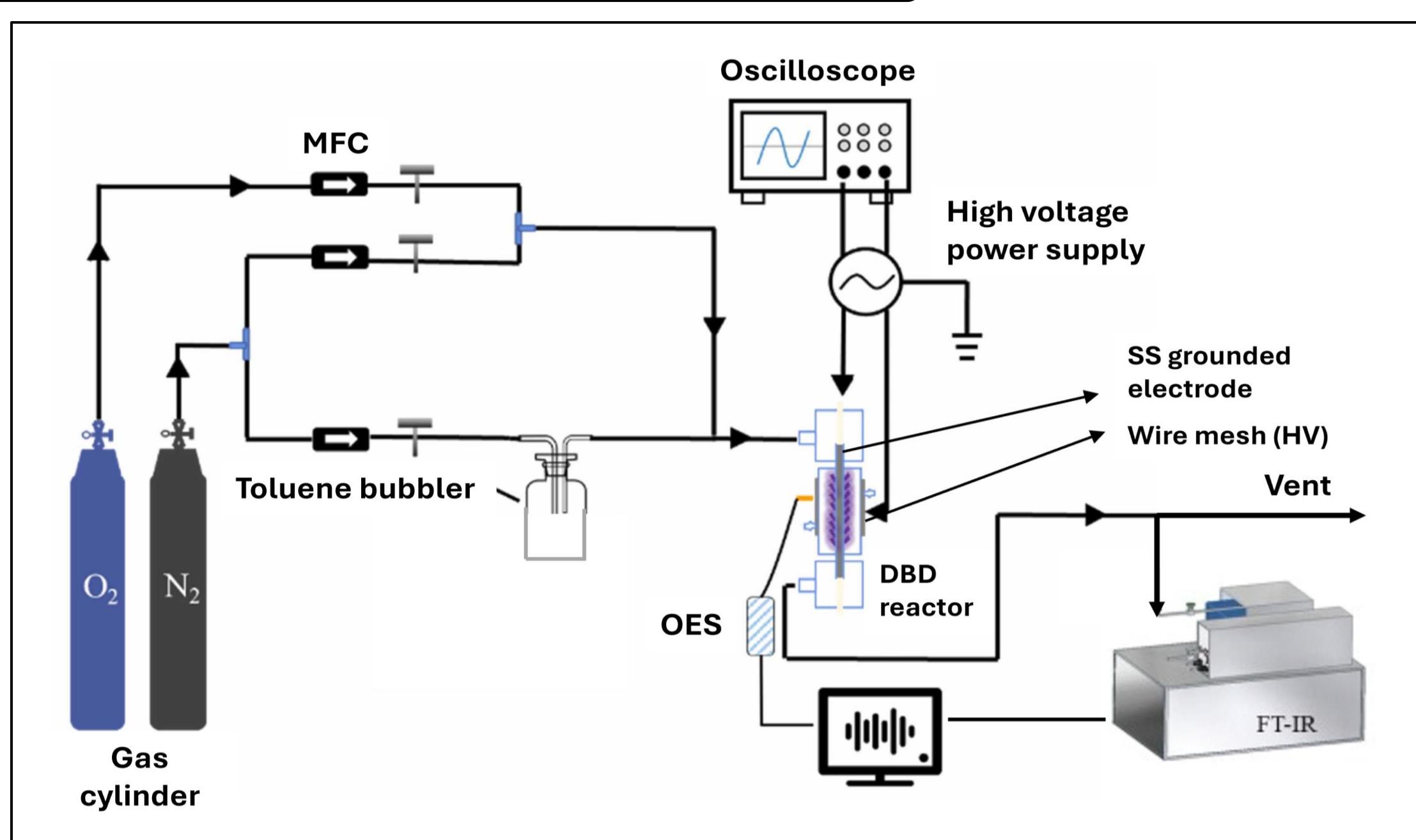
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Introduction and Experimental Setup

Plasma Reactor



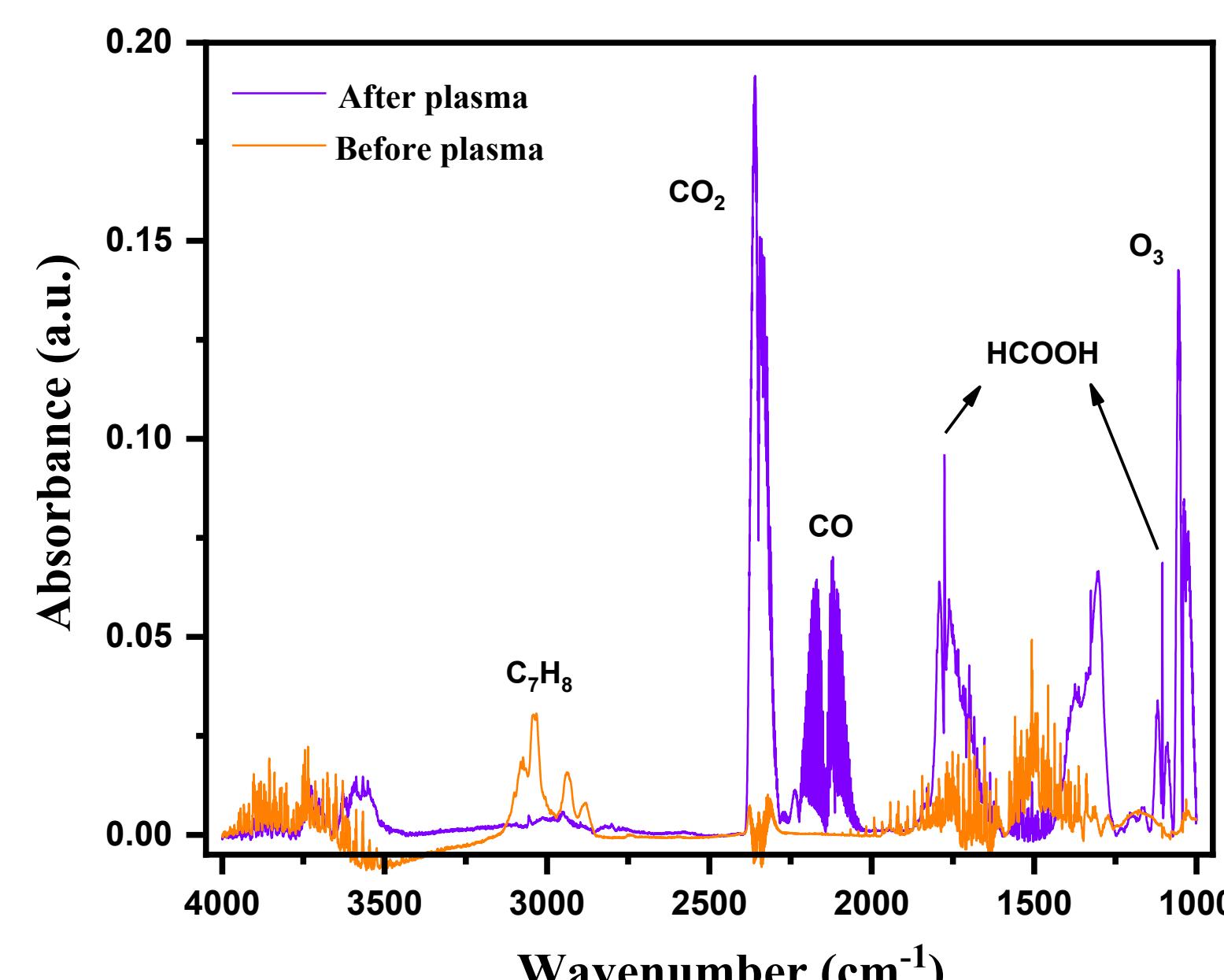
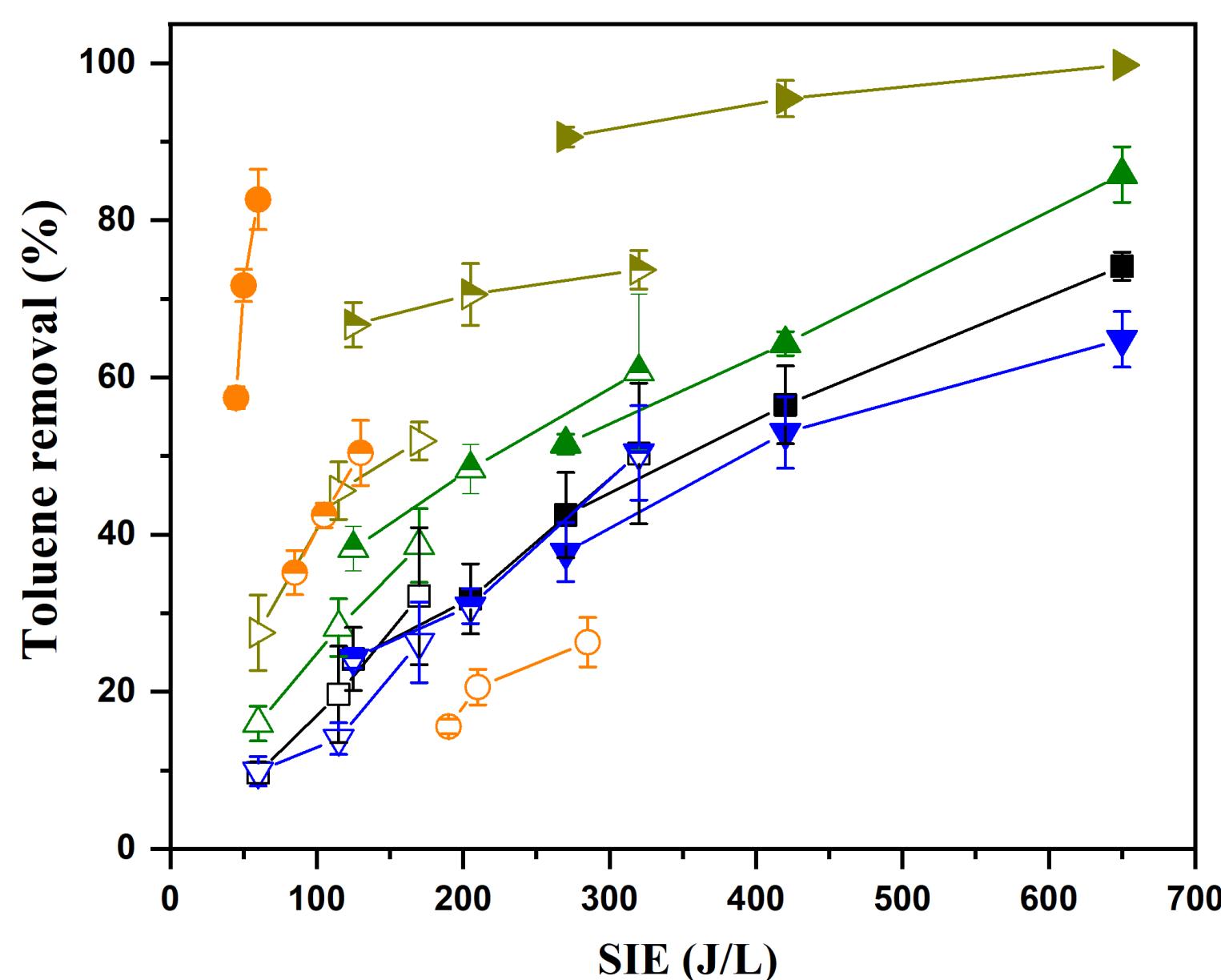
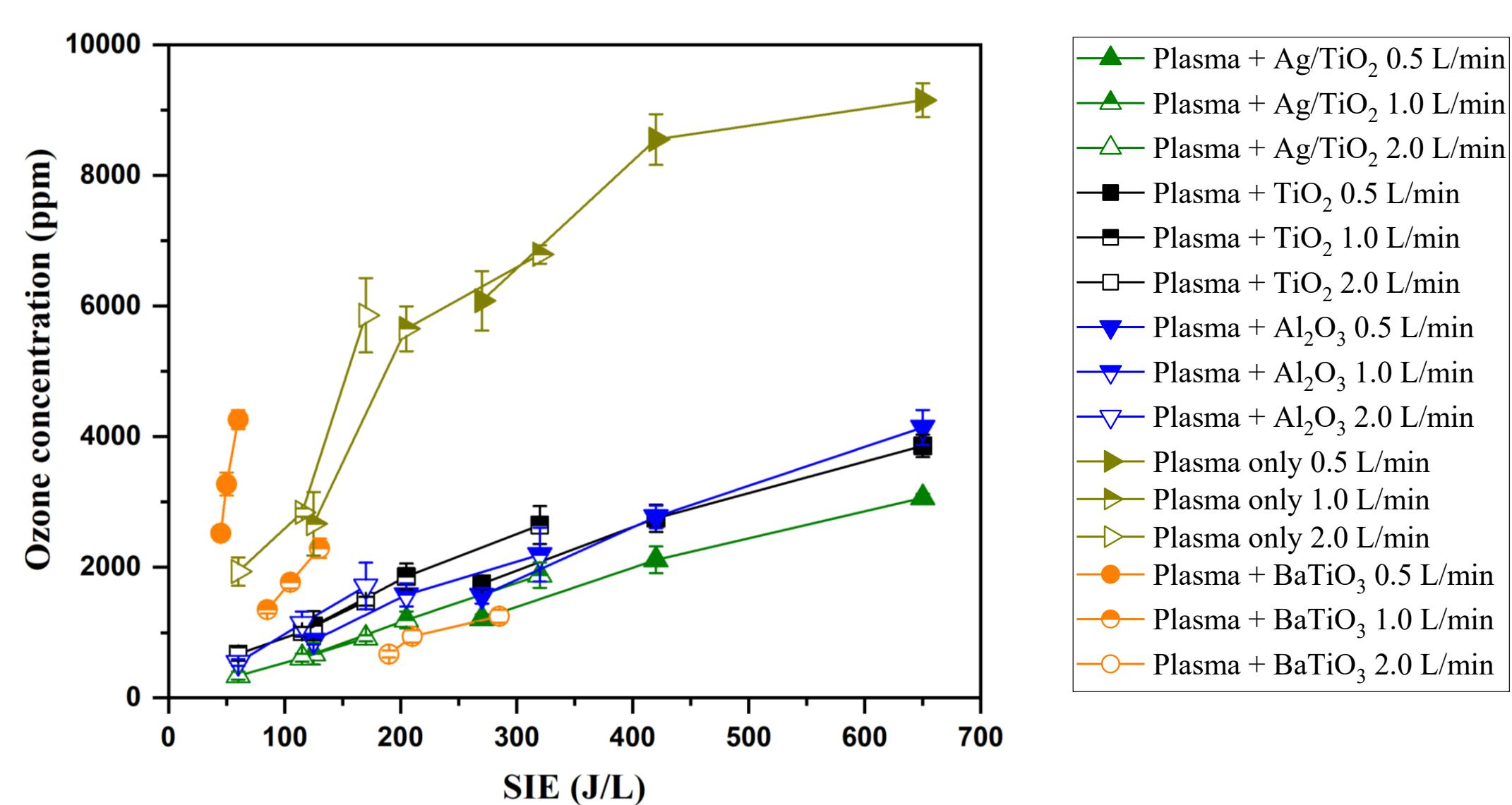
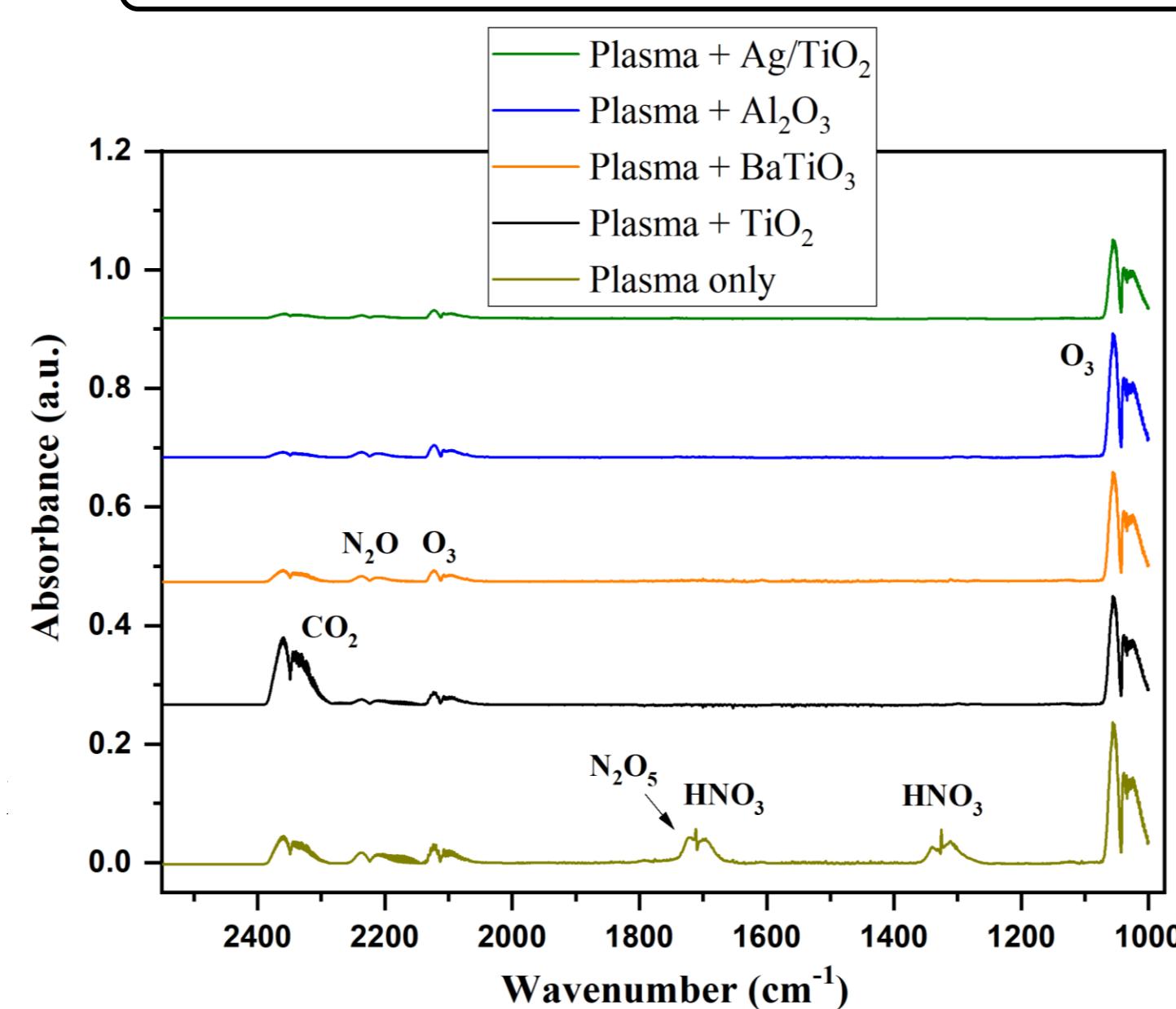
- Volatile organic compounds (VOCs) emissions from industries, vehicles, and consumer products are a major source of **air pollution**, contributing to smog, organic aerosols, and **ozone (O_3) formation**. Many VOCs (e.g., benzene, toluene, and formaldehyde) are **toxic or carcinogenic**, so their abatement is critical to protect public health and the environment.
- Non-thermal plasma can generate highly **reactive species** ($\cdot OH$, $O\cdot$, O_3 , NO_x radicals) enabling VOC degradation at **ambient temperature and pressure**. This makes plasma particularly attractive for treating VOC emissions where conventional methods (thermal incineration, adsorption, and biofiltration) are inefficient.
- Here, we conducted experiments using **different plasma reactors** filled with **catalytic pellets** and investigated their influence on **VOC degradation**. The reactive species generated in each reactor configuration were analyzed, and the **toluene removal efficiency** was compared across different plasma–catalyst combinations.



Plasma Reactor Characteristics and Toluene Degradation

Experimental Conditions

- Discharge reactor length: 120 mm
- Discharge (gas) gap: 3.5 mm
- Catalytic materials (pellets): $\gamma-Al_2O_3$, TiO_2 , $BaTiO_3$, Ag/TiO_2
- Carrier gas: N_2+O_2 (50:50)
- Gas flow rate: 0.5, 1.0, 2.0 L/min
- Applied voltage: 12 – 16.5 kV @ 1500
- Discharge power: 1 – 6 W



- For catalytic pellets with higher dielectric constant, a voltage drop across the dielectric is bigger than across the discharge gas gap. This reduces the discharge power delivered into the plasma, so the **SIE becomes lower**.
- The smaller voltage across discharge gas gap **weakens the electric field** in the plasma gap, reduces electron impact dissociation of O_2 , and therefore **O_3 production decreases**.
- Catalyst-assisted plasma enhances toluene removal efficiency, especially $BaTiO_3$ at lower **SIE**, highlighting the role of catalysts in improving VOC degradation.
- The presence of CO_2 , CO , $HCOOH$, and O_3 peaks after plasma treatment indicates **partial oxidation of toluene**, showing the need for optimized plasma–catalyst conditions.

- ✓ Plasma alone showed **higher energy efficiency** at lower **SIE**, but its efficiency decreased as **SIE** increased. In contrast, TiO_2 and $BaTiO_3$ catalysts maintained **more stable performance**, indicating their suitability for energy-efficient VOC abatement.



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