

# Non-Thermal Plasma-Assisted VOC Decomposition: Influence of Catalyst Type, Electron Energy, and Gas Residence Time

Ram Avtar Jangra, Gokul Selvaraj, Karol Hensel

Division of Environmental Physics, Faculty of Mathematics, Physics and Informatics,  
Comenius University Bratislava, Slovakia

Email: [avtar.ram@fmph.uniba.sk](mailto:avtar.ram@fmph.uniba.sk)

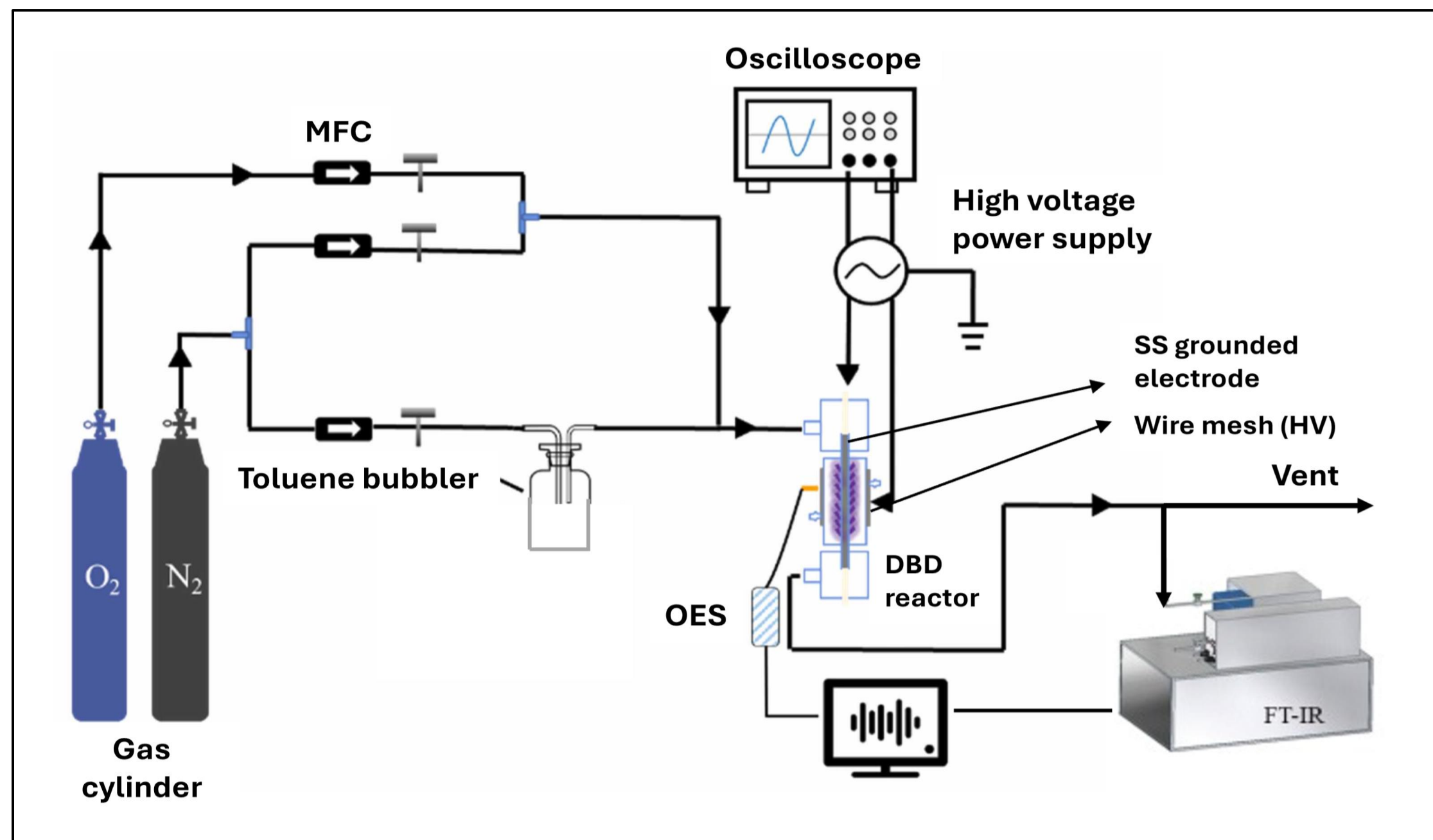


## Introduction and Experimental Setup

### Plasma Reactor



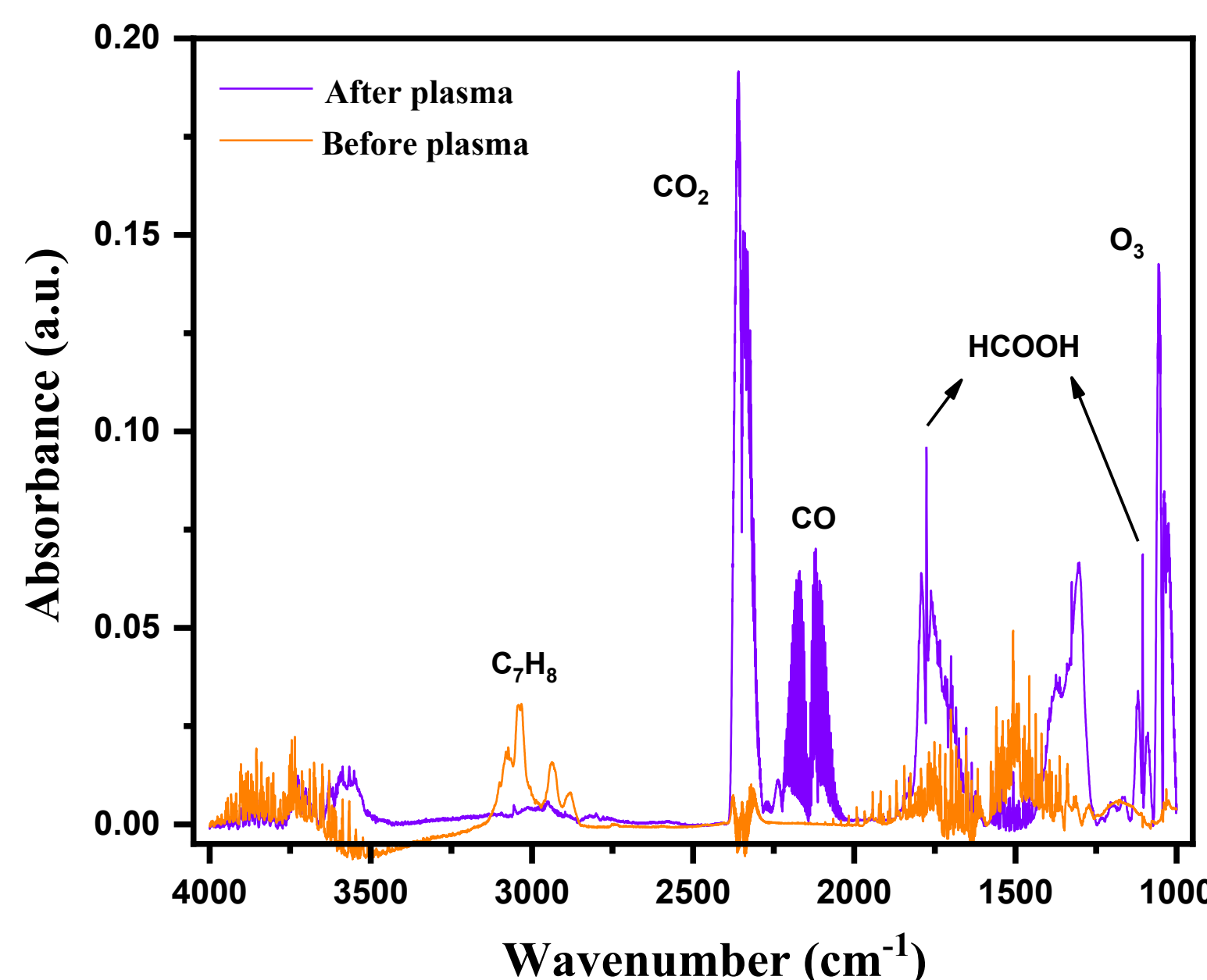
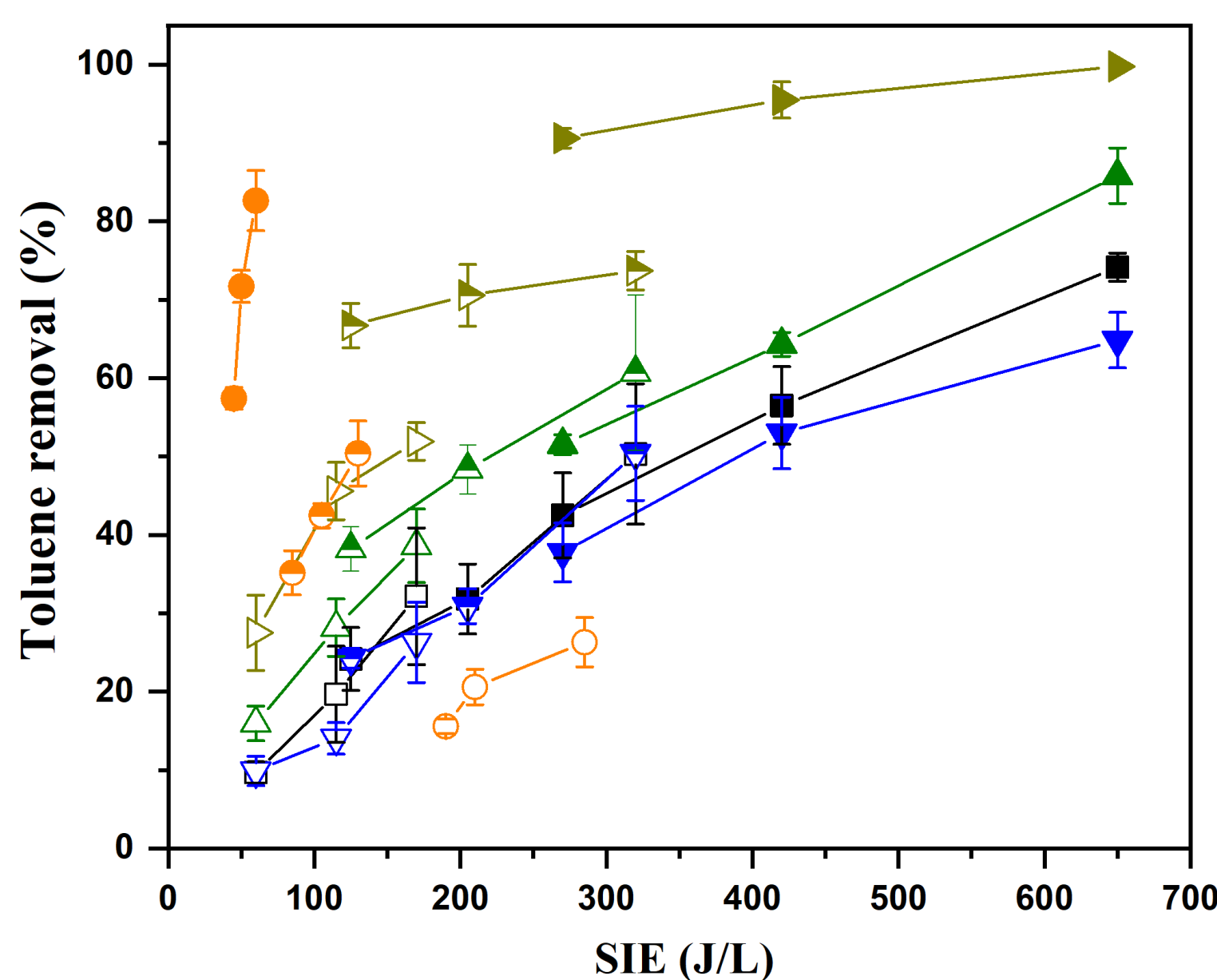
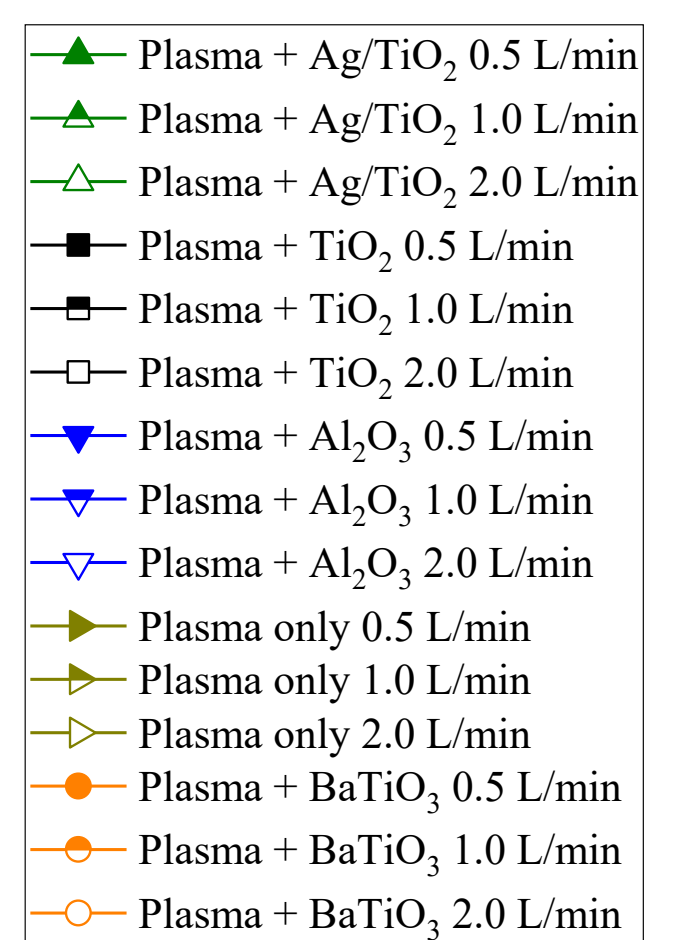
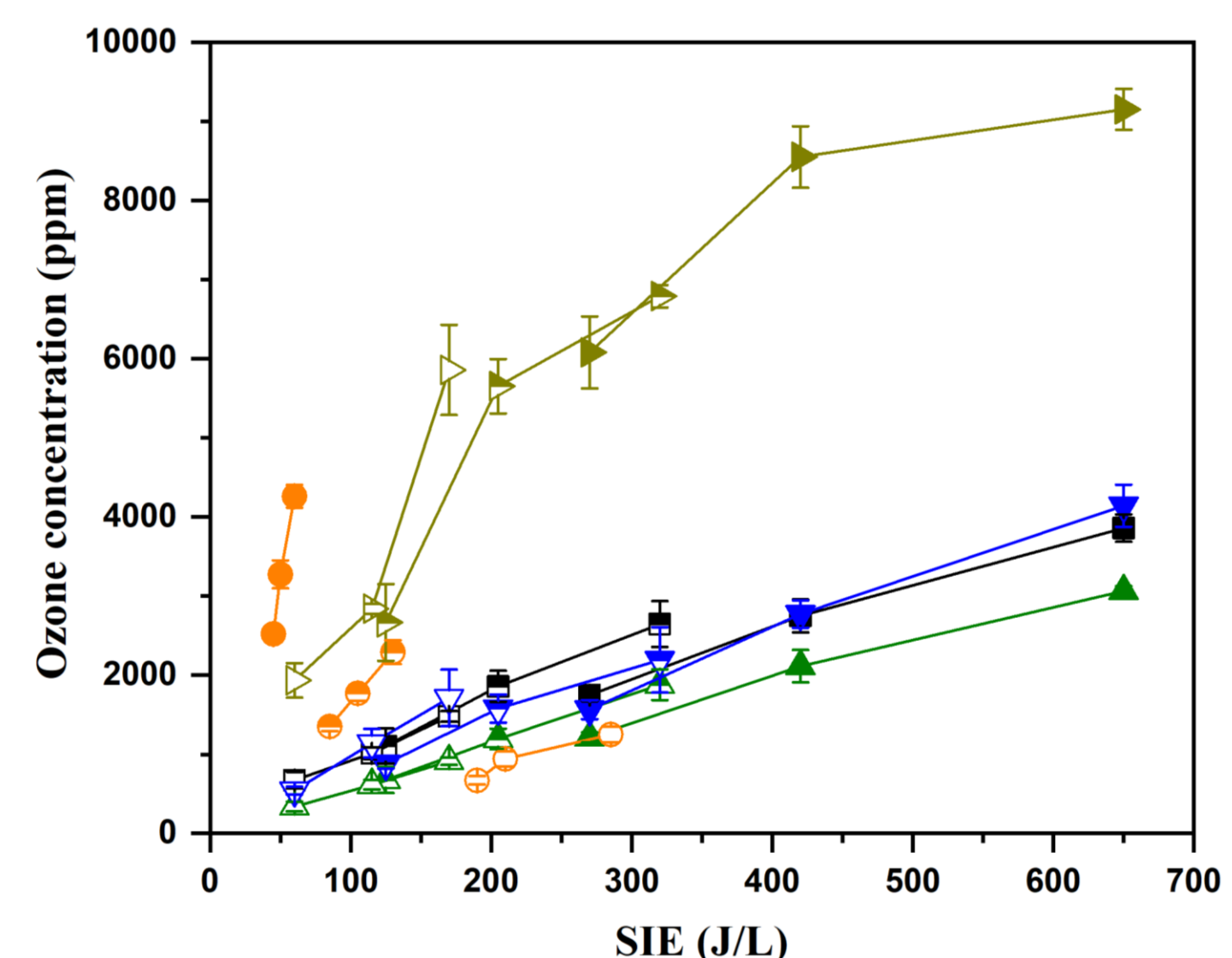
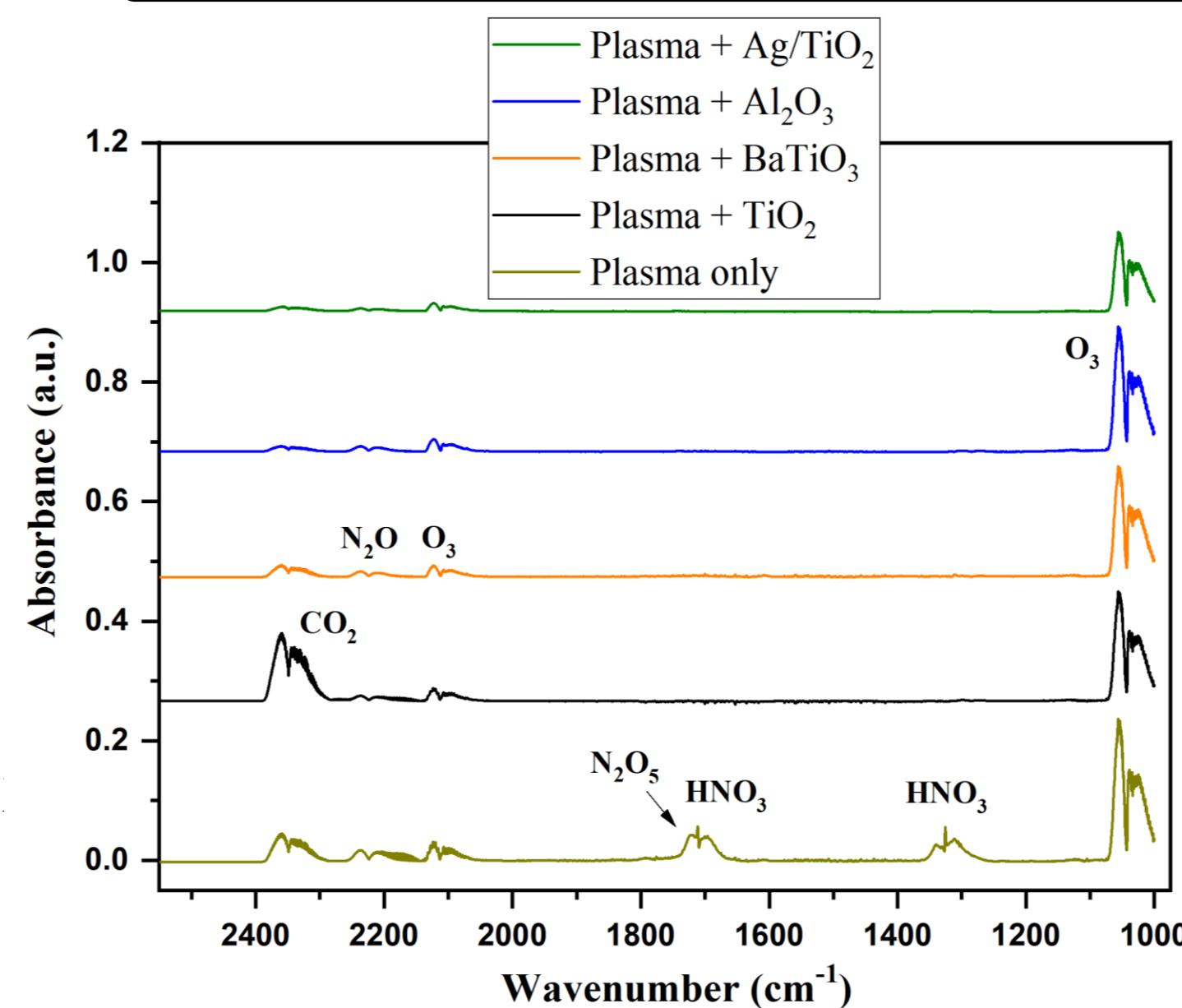
- ❑ **Volatile organic compounds (VOCs)** emissions from industries, vehicles, and consumer products are a major source of **air pollution**, contributing to smog, organic aerosols, and **ozone ( $O_3$ ) formation**. Many VOCs (e.g., benzene, toluene, and formaldehyde) are **toxic or carcinogenic**, so their abatement is critical to protect public health and the environment.
- ❑ **Non-thermal plasma** can generate highly **reactive species** ( $\cdot OH$ ,  $O\cdot$ ,  $O_3$ ,  $NO_x$  radicals) enabling VOC degradation at **ambient temperature and pressure**. This makes plasma particularly attractive for treating VOC emissions where conventional methods (thermal incineration, adsorption, and biofiltration) are inefficient.
- ❑ **Here**, we conducted experiments using **different plasma reactors filled with catalytic pellets** and investigated their influence on **VOC degradation**. The **reactive species** generated in each reactor configuration were analyzed, and the **toluene removal efficiency** was compared across different plasma-catalyst combinations.



## Plasma Reactor Characteristics and Toluene Degradation

### Experimental Conditions

- Discharge reactor length: 120 mm
- Discharge (gas) gap: 3.5 mm
- Catalytic materials (pellets):  $\gamma-Al_2O_3$ ,  $TiO_2$ ,  $BaTiO_3$ ,  $Ag/TiO_2$
- Carrier gas:  $N_2+O_2$  (50:50)
- Gas flow rate: 0.5, 1.0, 2.0 L/min
- Applied voltage: 12 – 16.5 kV @ 1500
- Discharge power: 1 – 6 W



- ❑ For catalytic pellets with higher dielectric constant, a voltage drop across the dielectric is bigger than across the discharge gas gap. This reduces the discharge power delivered into the plasma, so the **SIE becomes lower**.
- ❑ The smaller voltage across discharge gas gap **weakens the electric field** in the plasma gap, reduces electron impact dissociation of  $O_2$ , and therefore  **$O_3$  production decreases**.
- ❑ Catalyst-assisted plasma enhances toluene removal efficiency, especially  **$BaTiO_3$  at lower SIE**, highlighting the role of catalysts in **improving VOC degradation**.
- ❑ The presence of  $CO_2$ ,  $CO$ ,  $HCOOH$ , and  $O_3$  peaks after plasma treatment indicates **partial oxidation of toluene**, showing the need for optimized plasma-catalyst conditions.

✓ **Plasma alone** showed **higher energy efficiency at lower SIE**, but its efficiency decreased as SIE increased. In contrast,  **$TiO_2$  and  $BaTiO_3$  catalysts maintained more stable performance**, indicating their suitability for energy-efficient VOC abatement.