

Unraveling the Plasma–Photocatalysis Process: Coumarin Degradation Using a Modified Fountain DBD System

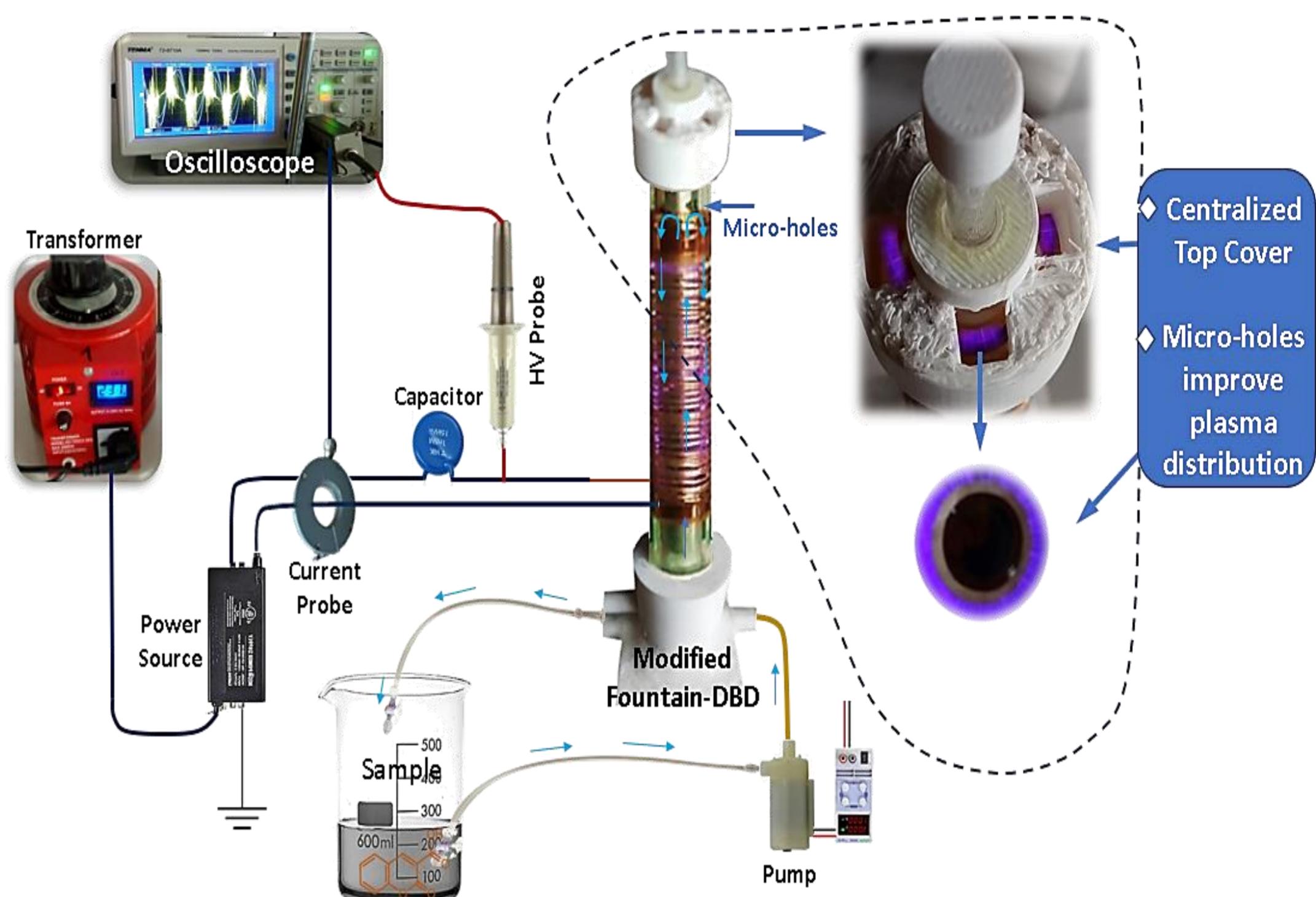
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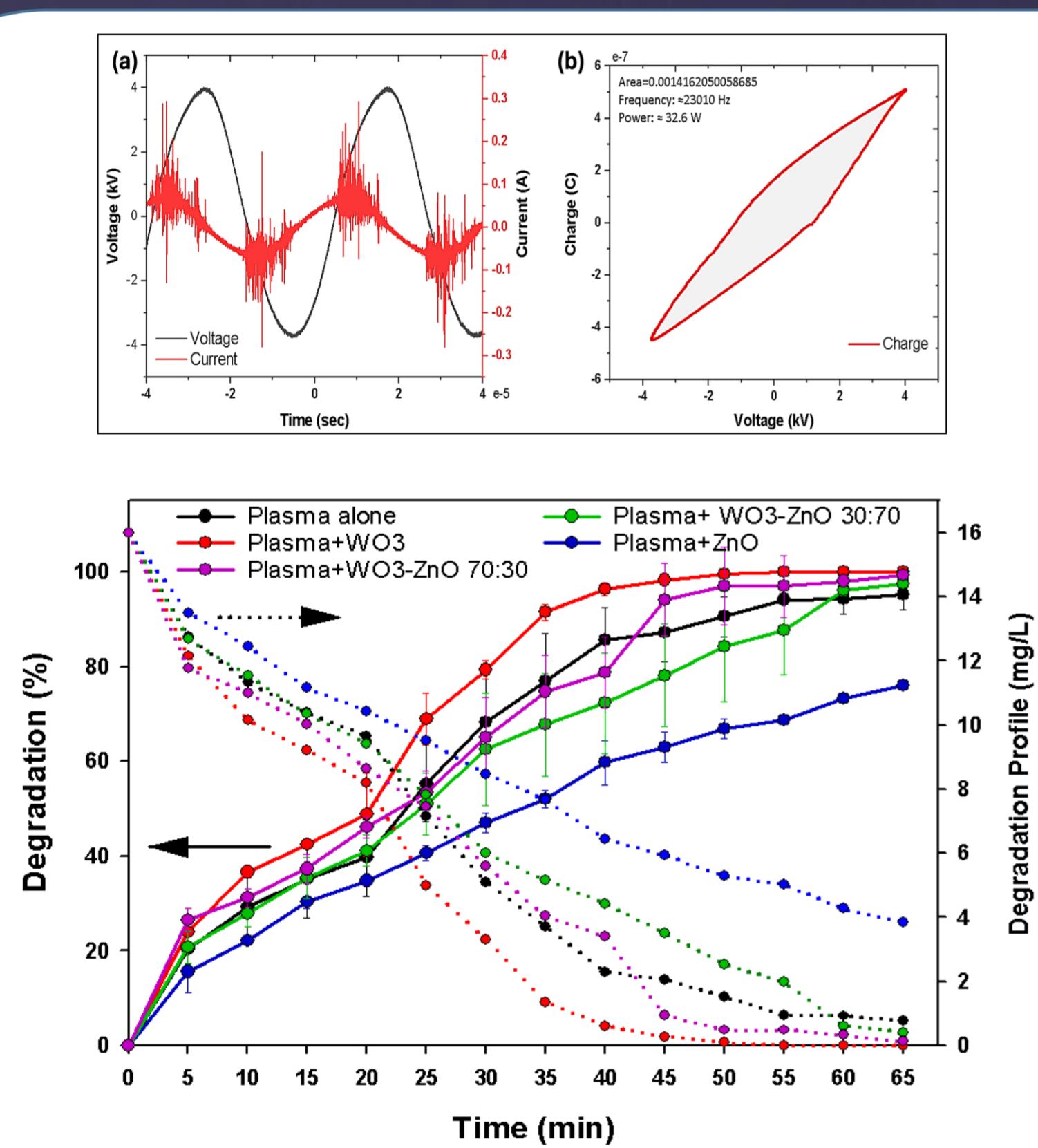
Abstract

Coumarin is widely used in cosmetics and pharmaceuticals, has emerged as a persistent and potentially toxic contaminant. This study investigates coumarin degradation using a modified fountain dielectric barrier discharge (MF-DBD) plasma reactor, operated alone and in combination with WO_3 , $\text{WO}_3\text{-ZnO}$, and ZnO catalysts. Under identical conditions (16 mg/L, atmospheric air pressure, 0–65 min), plasma treatment achieved 95% degradation, primarily via $\cdot\text{OH}$ -mediated oxidation. Incorporating catalysts significantly enhanced degradation kinetics and mineralization. WO_3 promoted reactive oxygen species (ROS) generation, while ZnO improved pH stability. The $\text{WO}_3\text{-ZnO}$ composites exhibited composition-dependent synergy, achieving >97% removal efficiency, with Plasma + WO_3 showing the highest performance (99.9% degradation, 84% TOC reduction). The results highlight the strong plasma–catalyst coupling effect and demonstrate MF-DBD plasma systems as efficient, green technologies for treating recalcitrant organic pollutants.

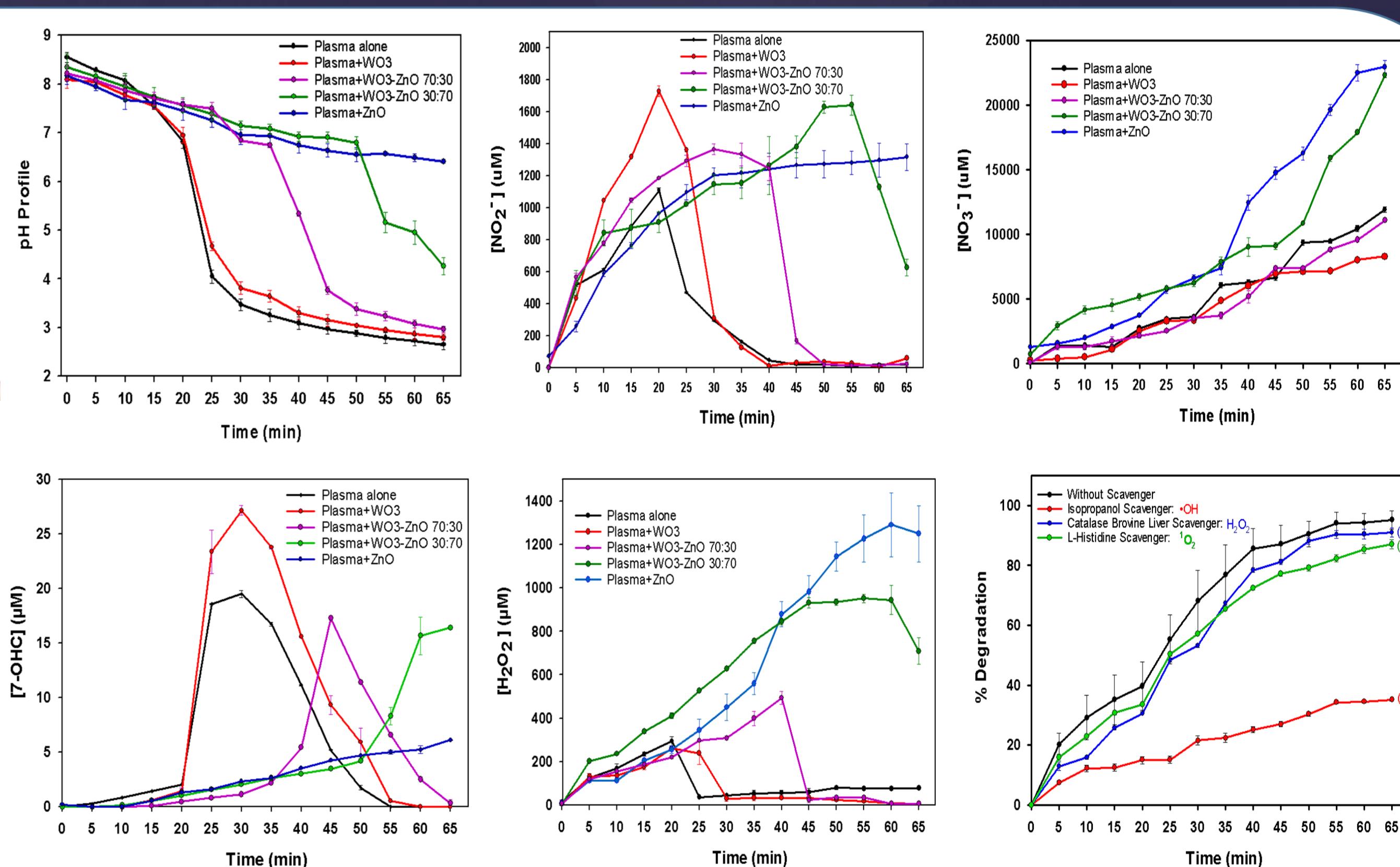
Experimental Set-Up



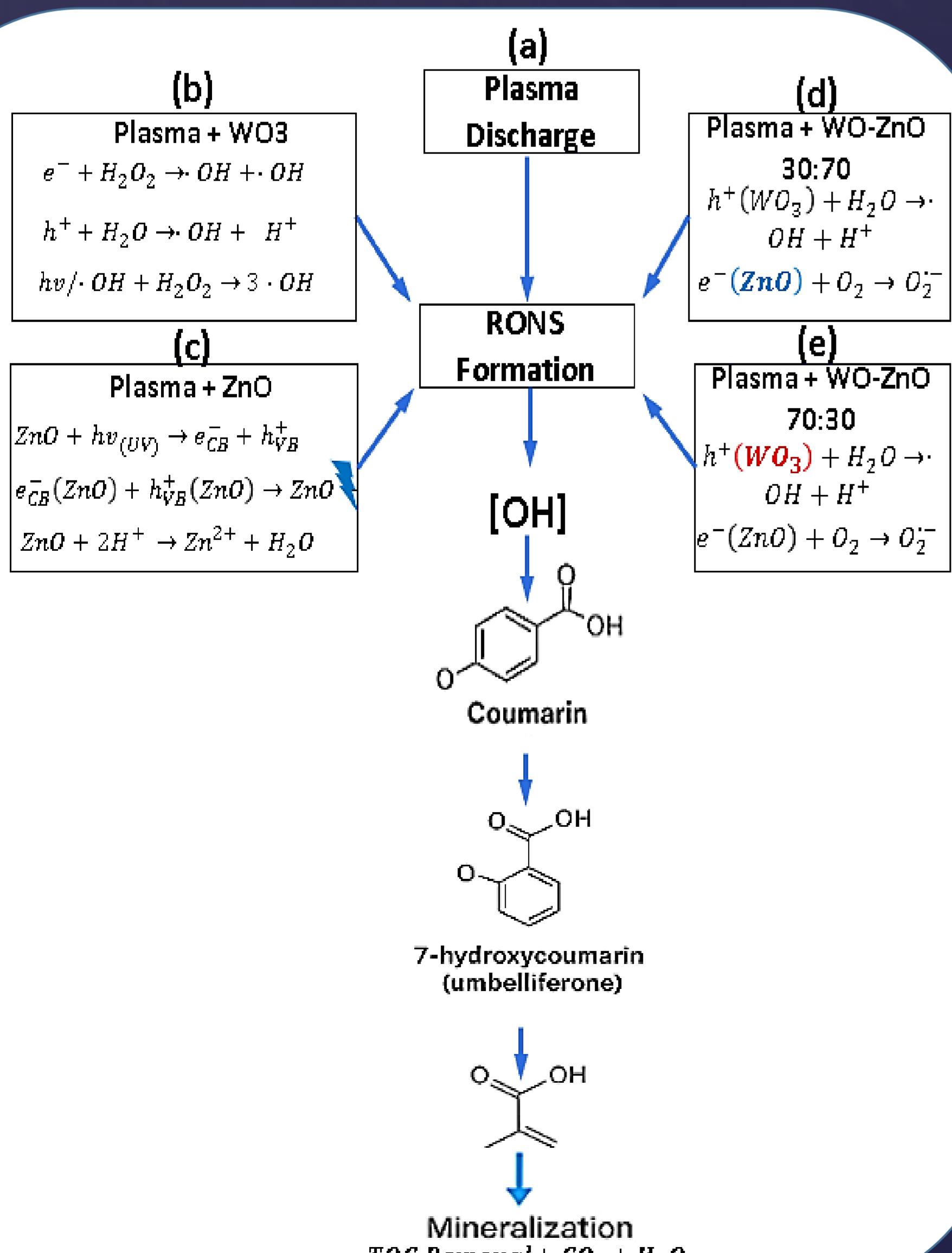
Results



Plasma Species



Proposed Pathway



Conclusion

- The MF-DBD plasma system effectively degraded coumarin under atmospheric conditions without additional oxidants.
- Catalyst coupling significantly enhanced degradation and mineralization through improved generation and utilization of reactive oxygen species.
- Among the tested systems, Plasma + WO_3 exhibited the highest efficiency (99.9% degradation, 84% TOC removal).
- The findings demonstrate that MF-DBD plasma–catalyst systems are a promising, sustainable technology for advanced treatment of persistent organic pollutants in water.

Acknowledgements

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