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Fluorescence emission of CO induced by electron impact

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Emission spectroscopy is particularly valuable in space exploration, as it enables remote analysis of planetary atmospheres, cometary comas, and nebulae irradiated mainly by nearby stars. In comets, atomic sulphur can be stored inside the dust grains or as part of volatiles in the form of OCS, H₂S, SO, S₂ or CS₂ [1, 2]. The comet's sulphur inventory is closely linked to the pre-solar cloud and holds important information about the degree of reprocessing of the material in the solar nebula and during comet accretion. Fluorescence emission spectra of CS₂ were obtained after electron impact at several electron energies. The spectrum measured at 100 eV shows a prominent emission from the CS₂⁺ ($\tilde{A}^2\Pi_u - \tilde{X}^2\Pi_g$) emission band. A few features of the CS₂⁺ ($\tilde{B}^2\Sigma_u^+ - \tilde{X}^2\Pi_g$) emission band were detected below 300 nm, and in the near-infrared region, the emission lines of CI, CII, SI, and SII were identified. Meanwhile, the emission spectrum measured at 10 eV shows solely the emission from neutral CS₂ ($1\Delta_u(V1B_2) - X^1\Sigma_g^+$) emission band, which spreads from 320-nm to 800-nm. There is no signal from ionized CS₂⁺, as the ionization energy of CS₂ is 10.073 eV [3].

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[2] D. Semenov et al., "Chemistry in disks: XI. Sulfur-bearing species as tracers of protoplanetary disk physics and chemistry: The DM Tau case," *Astron Astrophys*, vol. 617, Sep. 2018, doi: 10.1051/0004-6361/201832980.

[3] A. Kramida, Yu. Ralchenko, J. Reader, and NIST ASD Team, "NIST Atomic Spectra Database." Accessed: Feb. 20, 2024. <https://www.nist.gov/pml/atomic-spectra-database>.

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